

ISOLATION OF THE SESQUITERPENE ALCOHOL (–)-PACIFIGORGIOL FROM *VALERIANA OFFICINALIS*

R. BOS, H. HENDRIKS, J. KLOOSTERMAN* and G. SIPMA*

Laboratory of Pharmacognosy, State University of Groningen, Ant. Deusinglaan 2, NL-9713 AW Groningen, The Netherlands;

*Research Laboratory, PFW (Nederland) B.V., P.O. Box 3, NL-3800 AA Amersfoort, The Netherlands

(Received 27 August 1985)

Key Word Index—*Valeriana officinalis*; Valerianaceae; sesquiterpenoid alcohol; (–)-pacificorgiol.

Abstract—A tertiary bicyclic sesquiterpene alcohol has been isolated from *Valeriana officinalis*, and identified as (–)-pacificorgiol, the optical antipode of the alcohol isolated from *Pacifigorgia adamsii*.

In our work on sesquiterpenoids in the essential oil of *Valeriana officinalis* L. s. l. [1, 2] a compound was isolated by means of low pressure column chromatography and preparative TLC which proved to be a tertiary alcohol with a molecular formula $C_{15}H_{26}O$. The percentage of this compound in oils of different *V. officinalis* strains varied from 0.7 to 8.6% [Tittel, G. and Bos, R., unpublished results].

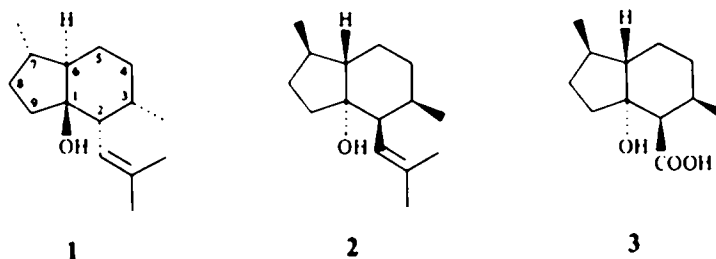
In the 100 MHz 1H NMR spectrum we identified four methyl groups at δ 0.75 (*d*, $J = 6.8$ Hz), 0.96 (*d*, $J = 6.2$ Hz), 1.60 (*d*, $J = 1.3$ Hz) and 1.73 (*d*, $J = 1.3$ Hz), a methine proton at 2.53 (*dd*, $J = 10.7$ resp. 4.3 Hz) and a vinylic proton at 5.06 [*m*, $J = 10.7$ Hz (*d*) and 1.3 Hz (*m*)]. The ^{13}C NMR spectrum showed four methyl groups (*q*) at δ 18.4, 19.0, 19.5 and 26.3; four CH_2 -groups (*t*) at δ 24.0, 29.4, 30.2 and 34.7; four CH -groups (*d*) at δ 30.3, 34.9, 47.7 and 49.2, carbon(s) of a tertiary alcohol group at δ 83.9; and a double bond at δ 134.4 (*s*) and 121.0 (*d*). The double bond was part of an isobutenyl group, $-CH=CHMe_2$, with chemical shifts at δ 5.06, 1.73 and 1.60 in 1H NMR, and δ 134.4, 121.0, 26.3 and 19.0 in ^{13}C NMR. There are three additional substituents, two methyl groups and a hydroxyl group, so the basic ring system is a bicyclononane system. A coupled ^{13}C NMR spectrum indicated that one of these rings was a cyclopentane ring,

with two CH_2 -groups and at least one CH -group. The ^{13}C NMR chemical shifts of the bridge carbon atoms correspond with those in a *trans*-bicyclo[4.3.0]nonane system [3], with a hydroxyl group in one of them.

From the chemical shifts on addition of increasing amounts of $Eu(fod)_3$ together with decoupling experiments, the most probable structure for this alcohol was 1. A similar compound, (+)-pacificorgiol (2) was found by Fenical *et al.* in a Pacific gorgonian coral *Pacifigorgia adamsii* [4], the structure of which was confirmed by 1H NMR (220 MHz) and X-ray analysis of the oxidation product 3. The published spectral data, IR and NMR, were identical to those of the alcohol from *Valeriana*, except for the optical rotation. (+)-Pacificorgiol isolated from the coral has an optical rotation $[\alpha]_D +41^\circ$ (*c* 1.02; $CHCl_3$), and the alcohol isolated from *Valeriana* with $[\alpha]_D -45.3^\circ$ (*c* 0.724; $CHCl_3$) will be named (–)-pacificorgiol.

EXPERIMENTAL

NMR spectra were recorded at 99.55 MHz for 1H and at 25 MHz for ^{13}C NMR. $CDCl_3$ was used as a solvent, with TMS as an internal standard. MS were recorded on a Finnigan 9500/3300/6110 GC/MS Computer system for EI-operation; a



fused silica capillary column (25 m \times 0.32 mm CP sil 5; Chrompack, Middelburg, The Netherlands) was used; temp. program 60–300° at 6°/min; carrier gas helium; 70 eV and a cycle time during acquisition of mass spectra: 1 sec and by direct inlet procedure.

Acknowledgements—The authors thank Miss F. M. S. v. Putten for technical assistance and Mr. A. J. Ottens (V.N.K., P.O. Box 1, NL-8080 AA Elburg) for *Valeriana officinalis*.

REFERENCES

1. Hendriks, H. and Bos, R. (1984) *Dragoco Report* 3.
2. Bos, R., Hendriks, H., Bruins, A. P., Kloosterman, J. and Sipma, G. (1986) *Phytochemistry* **25**, 133.
3. Pouet, M. J. (1982) *Org. Magn. Reson.* **19**, 229.
4. Izac, R. R., Poet, S. E., Fenical, D., Van Engen, D. and Clardy, J. (1982) *Tetrahedron Letters* **23**, 3743.

Phytochemistry, Vol. 25, No. 5, pp 1235–1237, 1986.
Printed in Great Britain.

0031-9422/86 \$3.00 + 0.00
© 1986 Pergamon Press Ltd

THE MICROBIOLOGICAL TRANSFORMATION OF CANDIDIOL, *ENT*-15 β ,18-DIHYDROXY-KAUR-16-ENE, BY *GIBBERELLA FUJIKUROI*

BRAULIO M. FRAGA, PEDRO GONZÁLEZ*, MELCHOR G. HERNÁNDEZ, FERNANDO G. TELLADO and AUREA PERALEST

Instituto de Productos Naturales Orgánicos, C.S.I.C., La Laguna, Tenerife, Canary Islands, Spain; *Departamento de Química Orgánica, Universidad de La Laguna, Tenerife, Spain; †Departamento de Rayos X, Instituto Rocasolano, C.S.I.C., Serrano 119, 28006-Madrid, Spain

(Revised received 23 September 1985)

Key Word Index—*Gibberella fujikuroi*; microbiological transformation; diterpenes; candidiol; *ent*-kaur-16-ene derivatives.

Abstract—Incubation of candidiol, *ent*-15 β ,18-dihydroxy-kaur-16-ene, with *Gibberella fujikuroi* affords *ent*-11 α ,15 β ,18-trihydroxy-kaur-16-ene. Its structure was determined by X-ray analysis.

The fungus *Gibberella fujikuroi* produces a series of metabolites such as the gibberellins and the kaurenolide lactones, which derive biogenetically from the diterpene *ent*-kaur-16-ene [1]. This fungus is able to handle some natural and synthetic diterpenoids. Thus the microbiological transformation of *ent*-kaur-16-ene derivatives hydroxylated at different positions has been carried out [2]. In contrast to these investigations the metabolism of 15 α -hydroxylated *ent*-kaurenes has been little studied. *ent*-15 β -Hydroxy-kaur-16-en-19-oic acid has been incubated with *G. fujikuroi* and the products obtained tentatively identified as 7 β ,15 α -dihydroxykaurenolide, GA₁₄ 7,15-lactone and a hydroxy GA₁₄ 7,15-lactone [3]. We now describe the results of incubation with this fungus of candidiol, *ent*-15 β ,18-dihydroxy-kaur-16-ene (1), a diterpene isolated from *Sideritis candicans* [4], and also synthesized from *ent*-18-hydroxy-kaur-15-ene [5].

The fermentation was carried out in the presence of AMO 1618, a compound that inhibits the formation of *ent*-kaur-16-ene without perturbing the post-kaurene metabolism [6, 7], thus facilitating the analysis of the products formed. Candidiol (1) was incubated with *G. fujikuroi*, the fermentation harvested after 6 days, and

the combined broth and mycelium extracts separated into neutral and acidic fractions.

Chromatography of the neutral fraction afforded a triol, C₂₀H₃₂O₃ (3). Its ¹H NMR spectrum showed resonances for two methyl groups and one hydroxymethylene group, the two protons of an exocyclic double bond and the geminal hydrogens to two secondary hydroxyl groups. This spectrum was very similar to that of candidiol (1) [4, 5], the difference being the appearance of a geminal hydrogen to a new alcoholic group. This compound formed a triacetate (4). Its ¹H NMR spectrum in CDCl₃ showed the geminal proton to one of the acetates overlapped with one of the olefinic hydrogens, but in C₆D₆ a doublet centred at δ 5.13 was observed. The form of this signal indicated that C-14 was a possible place for this new function, but the ¹³C NMR spectrum was more in accordance with an alcoholic group at C-11 (Table 1). To solve this ambiguity we submitted the alcohol to an X-ray analysis. In this way the structure of *ent*-11 α ,15 β ,18-trihydroxy-kaur-16-ene was assigned to this new metabolite.

The crystal structure of 3 consists of discrete molecules, one of which is shown in Fig. 1. Rings A, B and C have